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HASIL PENILAIAN SEJAWAT SEBIDANG ATAU *PEER REVIEW*
KARYA ILMIAH : JURNAL ILMIAH

Judul Jurnal Ilmiah (Artikel) : Spectrochemical analysis of Cs in water and soil using low pressure laser induced breakdown spectroscopy

Nama/ Jumlah Penulis : 5 Orang

Status Pengusul : ~~Penulis pertama~~ / Penulis ke 2 / ~~Penulis Korespondensi~~

Identitas Jurnal Ilmiah :

- a. Nama Jurnal : Spectrochimica Acta Part B: Atomic Spectroscopy
- b. Nomor ISSN : 0584-8547
- c. Vol, No., Bln Thn : Vol. 132, No.-, Maret 2017
- d. Penerbit : Elsevier
- e. DOI artikel (jika ada) : 10.1016/j.sab.2017.03.017
- f. Alamat web jurnal : <https://www.elsevier.com/locate/sab>
- Alamat Artikel : <https://www.sciencedirect.com/science/article/abs/pii/S0584854716302038>
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d. Kelengkapan unsur dan kualitas terbitan/jurnal (30%)	12			11,8
Total = (100%)	40			38,9
Nilai Pengusul = 20% x 38,9/4 = 1,95				

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Isi jurnal telah ditulis dengan memiliki kesesuaian antar bagian-bagiannya, gap riset jelas dan lengkap serta memenuhi unsur-unsur penulisan jurnal yang baik.

2. Ruang lingkup dan kedalaman pembahasan:

Aspek ruang lingkup jurnal telah sesuai, pembahasan telah dilakukan secara mendalam dengan hasil yang diperoleh

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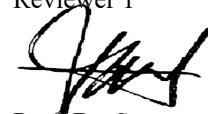
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Jurnal telah memiliki kelengkapan unsur yang sangat baik dan kualitas terbitan yang sangat baik, jurnal yang terindeks scopus, memiliki indek similaritas yang kecil.

Semarang, 20 April 2021

Reviewer 1



Prof. Dr. Suryono, S.Si., M.Si.

NIP. 197306301998021001

Unit Kerja : Fisika

Bidang Ilmu: Fakultas Sains dan Matematika

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d. Kelengkapan unsur dan kualitas terbitan/jurnal (30%)	12			11,5
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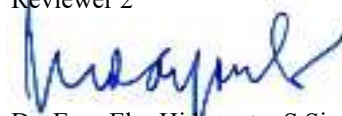
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4. Kelengkapan unsur dan kualitas terbitan:

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Semarang, 04 Juli 2021

Reviewer 2



Dr. Eng. Eko Hidayanto, S.Si., M.Si.

NIP. 197301031998021001

Unit Kerja : Fisika

Bidang Ilmu: Fakultas Sains dan Matematika

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b. Ruang lingkup dan kedalaman pembahasan (30%)	11,6	11,7	11,65
c. Kecukupan dan kemutakhiran data/informasi dan metodologi (30%)	11,7	11,7	11,7
d. Kelengkapan unsur dan kualitas penerbit (30%)	11,8	11,5	11,65
Total = (100%)			38,8
Nilai untuk Pengusul : $(20\% \times 38,8/4) = 1,94$			

Semarang, 24 Februari 2021

Reviewer 1



Prof. Dr. Suryono, S.Si., M.Si.
NIP. 197306301998021001

Bidang ilmu/Unit kerja : Fakultas Sains dan Matematika/Fisika

Reviewer 2



Dr. Eng. Eko Hidayanto, S.Si., M.Si.
NIP. 197301031998021001

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Spectrochemical analysis of Cs in water and soil using low pressure laser induced breakdown spectroscopy

Ramli M.^a, Khumaeni A.^b, Kurniawan K.H.^c ✉, Tjia M.O.^{c,d}, Kagawa K.^{c,e}

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^b Department of Physics, Faculty of Mathematics and Natural Sciences, Diponegoro University, Tembalang Campus, Semarang, Indonesia

^c Research Center of Maju Makmur Mandiri Foundation, 40/80 Srengseng Raya, Jakarta, 11630, Indonesia

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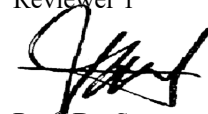
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Semarang, 20 April 2021

Reviewer 1



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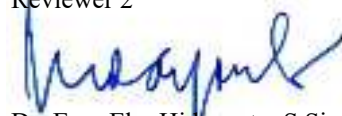
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Semarang, 04 Juli 2021

Reviewer 2



Dr. Eng. Eko Hidayanto, S.Si., M.Si.

NIP. 197301031998021001

Unit Kerja : Fisika

Bidang Ilmu: Fakultas Sains dan Matematika

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Laser-induced breakdown spectroscopy (LIBS), part II: Review of instrumental and methodological approaches to material analysis and applications to different fields
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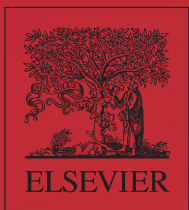
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Practical and highly sensitive elemental analysis for aqueous
samples containing metal impurities employing
electrodeposition on indium-tin oxide film samples and laser-
induced shock wave plasma in low-pressure helium gas
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(2015) *Applied Optics*, 54 (25), pp. 7592-7597. Cited 6 times.
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SPECTROCHIMICA ACTA

PART B: ATOMIC SPECTROSCOPY

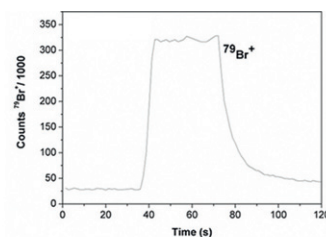
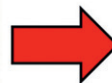
A comparison of laser ablation-inductively coupled plasma-mass spectrometry and high-resolution continuum source graphite furnace molecular absorption spectrometry for the direct determination of bromine in polymers

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Bromine determination



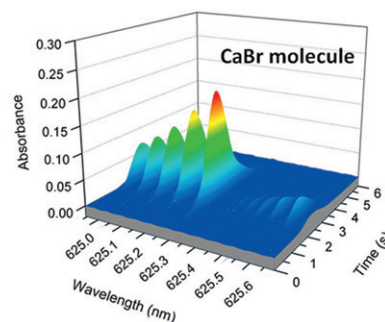
Plastic materials



LA-ICP-MS

X

HR-CS SS-GF MAS



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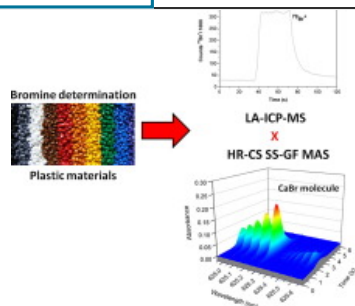
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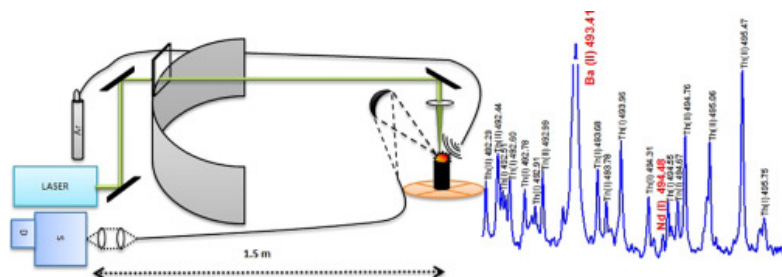
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Application of high-resolution continuum source flame atomic absorption spectrometry to reveal, evaluate and overcome certain spectral effects in Pb determination of unleaded gasoline

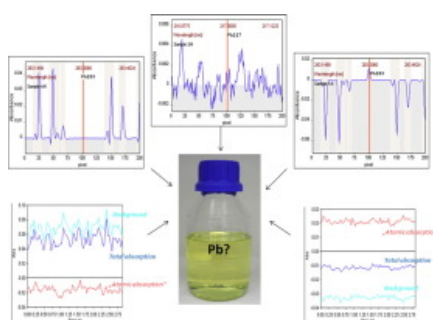
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Hideyuki Matsuta

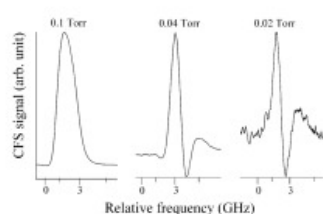
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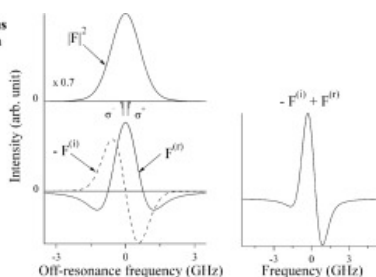
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Observed spectra and theoretically calculated Faraday functions of the 842.5 nm Ar I coherent forward scattering (CFS) spectra



CFS spectra of the 842.5 nm Ar I line at above partial pressures of Ar in a He-Ar RF glow discharge plasma. (see Fig.4 in the text)



Calculated profiles of the Faraday functions: $|F|^2$, $F^{(1)}$, $-F^{(1)}$, and $-F^{(1)} + F^{(0)}$. (see Fig. 7 in the text)

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Revealing hidden spectral information of chlorine and sulfur in data of a mobile Laser-induced Breakdown Spectroscopy system using chemometrics

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Abstract

Abstract

For the damage assessment of reinforced concrete structures the quantified ingress profiles of harmful species like chlorides, sulfates and alkali need to be determined. In order to provide on-site analysis of concrete a fast and reliable method is necessary. Low transition probabilities as well as the high ionization energies for chlorine and sulfur in the near-infrared range makes the detection of Cl I and S I in low concentrations a difficult task. For the on-site analysis a mobile LIBS-system ($\lambda = 1064$ nm, $E_{pulse} \leq 3$ mJ, $\tau = 1.5$ ns) with an automated scanner has been developed at BAM. Weak chlorine and sulfur signal intensities do not allow classical univariate analysis for process data derived from the mobile system. In order to improve the analytical performance multivariate analysis like PLS-R will be presented in this work. A comparison to standard univariate analysis will be carried out and results covering important parameters like detection and quantification limits (LOD, LOQ) as well as processing variances will be discussed (Allegrini and Olivieri, 2014 [1]; Ostra et

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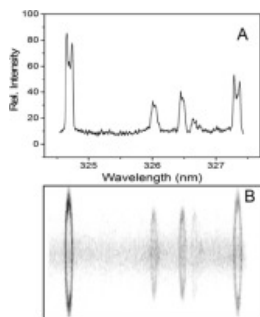
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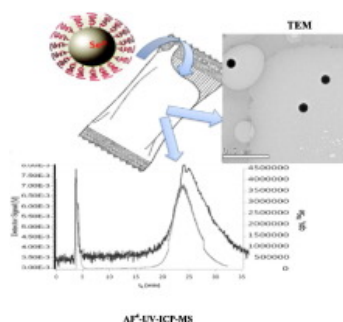
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María Palomo-Siguero, Paula Vera, Yolanda Echegoyen, Cristina Nerin, ... Yolanda Madrid

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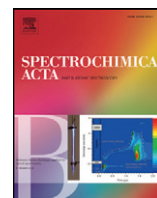
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Analysis of simulated high burnup nuclear fuel by laser induced breakdown spectroscopy

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ABSTRACT

Advanced Heavy Water Reactor (AHWR) grade (Th-U)₂O fuel sample and Simulated High Burn-Up Nuclear Fuels (SIMFUEL) samples mimicking the 28 and 43 GWd/Te irradiated burn-up fuel were studied using laser-induced breakdown spectroscopy (LIBS) setup in a simulated hot-cell environment from a distance of > 1.5 m. Resolution of < 38 pm has been used to record the complex spectra of the SIMFUEL samples. By using spectrum comparison and database matching > 60 emission lines of fission products was identified. Among them only a few emission lines were found to generate calibration curves. The study demonstrates the possibility to investigate impurities at concentrations around hundreds of ppm, rapidly at atmospheric pressure without any sample preparation. The results of Ba and Mo showed the advantage of LIBS analysis over traditional methods involving sample dissolution, which introduces possible elemental loss. Limits of detections (LOD) under Ar atmosphere shows significant improvement, which is shown to be due to the formation of stable plasma.

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1. Introduction

India constituting one-sixth of the world population is on the path of a rapid economic growth. Energy consumption being an essential part of development, per capita electricity consumption is considered as one of the index of civilization of a nation. With the rate of population growth and a bare minimum requirement of ~6000 kWh/year electrical consumption for acceptable quality of life in a society, India needs to improve its electricity generation nearly ten-folds in next four decades. Keeping this in mind, Department of Atomic Energy, India is pushing for utilisation of the vast Th resources of the country to produce electricity. Following this line of action, Advanced Heavy Water Reactor (AHWR) is being developed which will use (Th-²³³U)₂O fuel containing 3–3.75% U [1].

Elemental characterization of irradiated fuels is required for accessing the thermo-mechanical properties of the fuel in the post irradiation examination (PIE) stage. An unexpected value in any stage of irradiation can help to point out the possibility of a failure and the specific fuel bundle can be promptly removed. Another important reason behind elemental characterization during PIE is to know the fission yield of various fission products (FPs), which varies with the fission type (thermal or fast neutron fission) involved in the reactor and the fissile nuclei. These data also help the scientist to validate and modify reactor

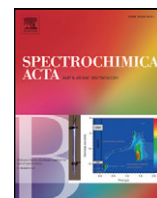
codes used by nuclear industry. Quantification of impurity constitutes is typically carried out by first dissolving irradiated fuels in acid solution, following bulk separation and subsequent analysis by radiometric or mass-spectrometric methods. Irrespective of the composition, the irradiated fuels due to their sintered nature are difficult to dissolve and is time consuming. It also poses high level radiation hazards, which is a driving force for developing desirable methods involving minimum sample handling procedures with an objective to minimize the radiation exposure to the workers.

Laser Induced Breakdown Spectroscopy (LIBS) being an emission spectroscopic method, has many advantages over conventional techniques, especially in the radioactive environment of nuclear industry. The notable advantages are, (i) LIBS can be a useful tool for in-situ remote analysis in hostile industrial environments at atmospheric pressure, since both the laser and emitted signal can be transmitted through optical fibres, (ii) possibility of analyzing both electrically conducting and non-conducting samples without the need of any conducting layer coating, (iii) possible to implement for fast and routine quality assurance checks during the fabrication stages, (iv) small sample requirement for the analysis, (v) micro analysis of inhomogeneous sample, and (vi) no sample preparation [2,3].

Detailed description of this technique is available in many reviews [4–6]. The possibilities of LIBS applications in the nuclear industries is reported by Wachter and Cremers in 1987 [7]. Uranium (U) has been studied by LIBS as a part of glass, reactor fuel, in solutions, in reduced pressure, as oxide, etc. [7–12]. Our group had carried out significant

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Application of high-resolution continuum source flame atomic absorption spectrometry to reveal, evaluate and overcome certain spectral effects in Pb determination of unleaded gasoline☆

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ABSTRACT

High-resolution continuum source and line source flame atomic absorption spectrometry (HR-CS FAAS and LS FAAS, respectively) were applied for Pb determination in unleaded aviation or automotive gasoline that was dissolved in methyl-isobutyl ketone. When using HR-CS FAAS, a structured background (BG) was registered in the vicinity of both the 217.001 nm and 283.306 nm Pb lines. In the first case, the BG, which could be attributed to absorption by the OH molecule, directly overlaps with the 217 nm line, but it is of relatively low intensity. For the 283 nm line, the structured BG occurs due to uncompensated absorption by OH molecules present in the flame. BG lines of relatively high intensity are situated at a large distance from the 283 nm line, which enables accurate analysis, not only when using simple variants of HR-CS FAAS but also for LS FAAS with a bandpass of 0.1 nm. The lines of the structured spectrum at 283 nm can have “absorption” (maxima) or “emission” (minima) character. The intensity of the OH spectra can significantly depend on the flame character and composition of the investigated organic solution. The best detection limit for the analytical procedure, which was 0.01 mg L^{-1} for Pb in the investigated solution, could be achieved using HR-CS FAAS with the 283 nm Pb line, 5 pixels for the analyte line measurement and iterative background correction (IBC). In this case, least squares background correction (LSBC) is not recommended. However, LSBC (available as the “permanent structures” option) would be recommended when using the 217 nm Pb line. In LS FAAS, an additional phenomenon related to the nature of the organic matrix (for example, isooctane or toluene) can play an important role. The effect is of continuous character and probably due to the simultaneous efficient correction of the continuous background (IBC) it is not observed in HR-CS FAAS. The fact that the effect does not depend on the flame character indicates that it is not radiation scattering. For LS FAAS, the determination of Pb using the 283 nm line, a 0.1 nm bandpass and a fuel lean flame is strongly recommended. The analysis of certified reference materials, recovery studies and the analysis of real samples with low Pb content supported the satisfactory accuracy of Pb determination in automotive or aviation gasoline when the recommended analytical variants are applied.

The studies in this work shed new light on spectral phenomena in air-acetylene flames. The structured background due to absorption by the OH molecules must be taken into account during Pb determination in other materials as well as in some other elemental determinations, especially at low absorbance levels. The usefulness of HR-CS FAAS for revealing and investigating a structured background was demonstrated. HR-CS FAAS does not reveal fully corrected spectral effects with a continuous character, which can be found in LS FAAS.

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1. Introduction

Lead is a very important element in the petroleum industry due to the application of tetraalkylleads (TALs) and especially the most widely

used compound TEL (tetraethyllead) as an antiknock agent [1–3]. TEL, which was commercially introduced in 1923, turned out to be an efficient and useful additive to gasoline. It was estimated that the application of this compound significantly contributed to the development of transport and the global economy [2]. However, TEL has also been called “the mistake of the 20th century” [2,3]. Its poisoning effect on human health, first of all, its neurotoxicity, has been known since the beginning of TEL production, but the facts were not properly taken into account for many years. The broad application of TALs contributed to significant environmental pollution all over the world, including in Greenland snow

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