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KARYA ILMIAH : JURNAL ILMIAH

Judul Jurnal Ilmiah (Artikel) : The Characterisation of MWCNT-rGO-TiO₂ Nanocomposite as Potential Electrode Material for Hybrid Supercapacitor

Nama/ Jumlah Penulis : 4 Orang

Status Pengusul : ~~Penulis pertama~~/Penulis ke 3/~~Penulis Korespondensi~~ **

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- a. Nama Jurnal : International Journal of Electrochemical Science
- b. Nomor ISSN : 1452-3981
- c. Vol, No., Bln Thn : Vol 17 No 5, Mei 2022 Hal 1-10
- d. Penerbit : Electrochemical Science Group
- e. DOI artikel (jika ada) : 10.20964/2022.05.10
- f. Alamat web jurnal : <http://www.electrochemsci.org/list22.htm#issue5>
- Alamat Artikel : www.electrochemsci.org/papers/vol17/22053.pdf
- g. Terindex : Scopus (Q4, SJR 0,31)

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Total = (100%)			38,35
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Semarang, 8 Maret 2023

Reviewer 1

Reviewer 2



Prof. Dr. Rahmat Gernowo, M.Si
 NIP. 196511231994031003
 Unit Kerja: FSM Universitas Diponegoro
 Bidang Ilmu: Fisika



Prof. Dr. Kusworo Adi, S.Si., M.T.
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Semarang, 1 Maret 2023
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Prof. Dr. Drs. Rahmat Gernowo, M.Si.
 NIP. 196511231994031003
 Unit Kerja : Fisika
 Bidang Ilmu: Fakultas Sains dan Matematika

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Prof. Dr. Kusworo Adi, S.Si., M.T.
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^a Physics Department, Science and Mathematics Faculty, Diponegoro University, Jl. Prof. Sudarto, SH, Tembalang, Semarang, 50275, Indonesia

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Abstract

Increasing energy demands are triggering the development of high performance, low-cost supercapacitor electrode material, particularly in countries that are shifting to renewable energy as their current energy supply cannot meet their significant energy demands. These countries require high-voltage, safe and long-lasting electrical energy storage devices. Multi-walled carbon nanotubes (MWCNTs), reduced graphene oxide (rGO) and Titanium Dioxide (TiO₂) were tested during an in-situ

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Highly Enhanced Photocatalytic Hydrogen Production Performance of Heterostructured Ti₃C₂/TiO₂/rGO Composites

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Cui, J. , Cao, L. , Zeng, D. (2018) *Scanning*

High-performing and ultra-stable TiO₂ nanospheres as electrode materials for pseudo-supercapacitors

Dhilip Kumar, R. , Balachandran, S. , Jagadeesh Kumar, A. (2023) *Materials Letters*

Facile synthesis of nanosphere like rare-earth/transition metal dual-doped TiO₂ nanostructure for application as supercapacitor electrodes material

Munawar, T. , Manzoor, S. , Mukhtar, F. (2022) *Journal of Materials Science*

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hydrothermal process. The structure test of X-ray diffraction (XRD) and scanning electron microscope (SEM) characterisation was used to determine the relation between the morphology of the electrode and its electrochemical value. The morphology of the composite electrode indicated that the nanocomposite electrode had better values than the single material electrode. Additionally, cyclic voltammetry (CV) and electrochemical impedance spectrum (EIS) tested in a two-electrode system indicated that the specific capacitance of MWCNT-rGO-TiO₂ 30% was 168 F g⁻¹ and showed the energy and power density to be 15 Wh kg⁻¹ and 337.5 W kg⁻¹, respectively. This result was higher with the other electrode due to the faradaic and non-faradaic processes in the electrode. © 2022. The Authors. Published by ESG. All Rights Reserved.

Author keywords

electrochemical reaction; hydrothermal method; morphology; MWCNT; rGO; supercapacitor; TiO₂

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Voltammetric Behaviour of an Electroactive of Zn^{2+} - octa(aminopropyl)silsesquioxane nanocomposite : An Application to 4-Chlorophenol Detection in Tap Water

Fernanda dos Santos Franco¹, Murilo Santos Peixoto¹, Abner Santos Baroni¹, Alessandro dos Santos Felipe², Priscila Fernanda Pereira Barbosa¹, Vitor Alexandre Maraldi¹, Rebeca Moreira Lima Freitas², Stefani Fernanda Vicente Da Silva¹, Vanessa Petronília Alves², Alvaro Adriano Couto Moraes², Idalci Cruvinel dos Reis², Devaney Ribeiro do Carmo^{1,*}

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This paper presents the voltammetric behavior of electroactive metallosilsesquioxane and their use in the detection of 4-CP. The metallosilsesquioxane was synthesized in two steps. First by direct reaction between hydrochloride octa(aminopropyl)silsesquioxane ($\text{POSS-NH}_3^+\text{Cl}^-$) and Zn^{2+} a second by electrostatic interaction with $[\text{FeCN}_6]^{4-}$ (SZnFe). The nanostructured material showed a well-defined three redox couples with formal potential (E^0) of approximately 0.33, 0.78 and $1.01 \pm 0.01\text{V}$ (vs Ag/AgCl). Voltammetric analysis showed that the formal potential (E^0) strongly depends on concentration and electrolyte type, but the system remained unchanged with pH values between 2 and 8. The SZnH graphite paste electrode exhibits a sensitive and catalytic oxidation response for the detection of 4-CP. For detection of 4-CP, the modified graphite paste electrode had a linear range of 6.0×10^{-7} to $6.0 \times 10^{-5} \text{ mol L}^{-1}$ with a limit of detection of $5.3 \times 10^{-6} \text{ mol L}^{-1}$ and a sensitivity of $0.53 \text{ A/mol L}^{-1}$. This electrode could be used to detect 4-CP in tap water.

Keywords: Metallosilsesquioxane; Octa(aminopropyl)silsesquioxane; Prussian blue analogs; Voltammetry; Graphite Paste Electrode; 4-Chlorophenol.

1. INTRODUCTION

Research in advanced polymers, composites and nanostructured materials has collaborated to the development of new materials for technological application in recent years, constituting a research center that attracts researchers from other fields of knowledge.

Corrosion Behavior of Zinc-Aluminum-Magnesium Coated Steel in Simulated Marine Atmosphere

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The corrosion behavior of zinc-aluminum-magnesium (ZAM) coated steel in a simulated marine atmosphere was investigated using compositional, morphological, electrochemical, and mass loss analyses. The results revealed that the corrosion rate of ZAM coated steel fluctuated because of the corrosion products formed during its exposure to the marine atmosphere, and the mass loss after 1848 h of exposure was $27.37 \text{ g} \cdot \text{m}^{-2}$ after 1848 hours' exposure. The corrosion preferentially occurred in the eutectic phases, especially in the eutectic phases near the primary zinc phase and with an area smaller than the primary zinc phase. As the exposure time increased, some of the corrosion products flaked off from the coating, leading to the aggravation of local corrosion and the formation of corrosion pits. Moreover, the corrosion products in the corrosion pits were dissolved, which can be attributed to the decrease in pH caused by the consumption of hydroxides during the chemical reactions of the corrosion process.

Keywords: ZAM coated steel; atmospheric corrosion; mass loss; electrochemical measurements; corrosion pits

1. INTRODUCTION

The structural steel coated by galvanized coatings is widely used for home appliance and automotive applications. The elements of Al, Mg, Si were added to coatings in order to increase corrosion resistance and satisfy the requirements of manufacturing. ZAM coating, a novel coating developed from galvanized coating, has an excellent corrosion resistance and is being used in buildings,