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HASIL PENILAIAN SEJAWAT SEBIDANG ATAU PEER REVIEW
KARYA ILMIAH: JURNAL ILMIAH

Judul Jurnal Ilmiah (Artikel) : Ultrafiltration Membrane for Degumming of Crude Palm Oil-Isopropanol Mixture
 Jumlah Penulis : 3 orang
 Status Pengusul : Penulis pertama/ penulis ke-3/ penulis korespondensi
 Identitas Jurnal Ilmiah :
 a. Nama Jurnal : Chemical and Biochemical Engineering Quarterly
 b. Nomor ISSN : 0352-9568
 c. Volume, nomor, bulan, tahun : Vol. 32 No. 3, Juli 2018
 d. Penerbit : Croatian Society of Chemical Engineers
 e. DOI Artikel : 10.15255/CABEQ.2017.1244
 f. Alamat web Jurnal : <http://silverstripe.fkit.hr/cabeg/>
 Alamat artikel : <http://silverstripe.fkit.hr/cabeg/assets/Uploads/05-3-2018.pdf>
 g. Terindeks : SCOPUS (Q2)

Kategori Publikasi Jurnal Ilmiah
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Hasil Penilaian *Peer Review*

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	Reviewer I	Reviewer II	
a. Kelengkapan unsur isi Artikel (10%)	4	4	
b. Ruang lingkup dan kedalaman pembahasan (30%)	11.2	10	
c. Kecukupan dan kemutakhiran data/ informasi dan metodologi (30%)	11.2	11	
d. Kelengkapan unsur dan kualitas terbitan/ jurnal (30%)	11.2	11	
Total = (100 %)	37.6	36	36.8
Nilai pengusul = (60% x 36.8) = 22.08			

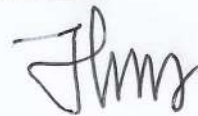
Semarang,

Reviewer 2



Prof. Dr. Andri Cahyo Kumoro, ST., MT.
 NIP. 197405231998021001
 Unit Kerja : Departemen Teknik Kimia FT UNDIP

Reviewer 1



Prof. Dr. Tutuk Djoko Kusworo, ST., M.Eng.
 NIP. 197306211997021001
 Unit Kerja : Departemen Teknik Kimia FT UNDIP

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Hasil Penilaian *Peer Review*

Komponen yang dinilai	Nilai Maksimal Jurnal Ilmiah			Nilai Akhir yang diperoleh
	Internasional	Nasional Terakreditasi	Nasional Tidak Terakreditasi	
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a. Kelengkapan unsur isi Artikel (10%)	4			4
b. Ruang lingkup dan kedalaman pembahasan (30%)	12			11.2
c. Kecukupan dan kemutakhiran data/informasi dan metodologi (30%)	12			11.2
d. Kelengkapan unsur dan kualitas terbitan / jurnal (30%)	12			11.2
Total = (100%)	40			37.6
Nilai pengusul = 60% x 37.2 = 22.3				

Catatan penilaian artikel oleh Reviewer:

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Makalah ini memiliki kelengkapan unsur sebagai makalah ilmiah yang bereputasi. Isis artikel sudah sesuai dengan scope atau bidang pada Jurnal Chemical and Biochemical Engineering Quarterly (10%)

2. Ruang lingkup dan kedalaman pembahasan:

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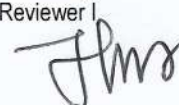
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Semarang,
 Reviewer I



Prof. Dr. Ir. Tutuk Djoko Kusworo, ST., M.Eng.
 NIP. 197306211997021001

Unit Kerja : Departemen Teknik Kimia FT UNDIP

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Komponen yang dinilai	Nilai Maksimal Jurnal Ilmiah			Nilai Akhir yang diperoleh
	Internasional	Nasional Terakreditasi	Nasional Tidak Terakreditasi	
	40	<input type="text"/>	<input type="text"/>	
a. Kelengkapan unsur isi artikel (10%)	4			4
b. Ruang lingkup dan kedalaman pembahasan (30%)	12			10
c. Kecukupan dan kemutakhiran data/informasi dan metodologi (30%)	12			11
d. Kelengkapan unsur dan kualitas terbitan/ jurnal (30%)	12			11
Total = (100%)	40			36
Nilai pengusul = 60% x 36 = 21,6				

Catatan penilaian artikel oleh Reviewer:

- Kesesuaian dan kelengkapan unsur isi artikel:** Artikel ini mencakup judul, abstrak, metode percobaan, hasil dan pembahasan, kesimpulan, ucapan terima kasih dan daftar pustaka.
- Ruang lingkup dan kedalaman pembahasan:** Penelitian dimaksudkan untuk memisahkan gum dan fosfolipid dari minyak sawit mentah dan campurannya dengan isopropanol menggunakan membrane sebagai metode alternatif metode pemisahan yang hemat energi. Beberapa model fisika terkait dengan penerapan system membrane (fluks, rejeksi dan fouling) juga diujikan. Hasil percobaan dibahas dengan cukup jelas secara teoritik dan diskriptif dengan dukungan pustaka yang memadai walaupun hanya 29,55% dr semua pustaka yang digunakan (44 buah).
- Kecukupan dan kemutakhiran data/informasi dan metodologi:** Latar belakang, metodologi serta hasil penelitian disajikan dengan cukup baik. Namun, dari 44 pustaka yang digunakan hanya 30% pustaka yang bersifat mutakhir (10 th terakhir).
- Kelengkapan unsur dan kualitas terbitan:** Journal ini tergolong dalam jurnal internasional bereputasi Q3 (process chemistry and technology) dengan SJR 0.358 dan H index 33 pada pemeringkatan jurnal Scimago tahun 2018 (pada saat artikel terbit). Editorial board berasal dari berbagai negara, terdapat pedoman penulisan yang jelas dan format penulisan konsisten. Pada umumnya, artikel yang disajikan merupakan hasil karya penulis dari berbagai negara. Penerbit jurnal ini adalah Croatian Society of Chemical Engineers. Gaya selingkung yang diterapkan jurnal ini sangat baik. Sebanyak 10 artikel yg terbit pada volume tsb ditulis oleh penulis yg berasal dari 9 negara.

Semarang, 15 Januari 2020
Reviewer 2


 Prof. Dr. Andri Cahyo Kumoro, ST., M.T.
 NIP. 197405231998021001
 Unit Kerja: Departemen Teknik Kimia FT UNDIP



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Chemical and Biochemical Engineering Quarterly [Open Access](#)
Volume 32, Issue 3, October 2018, Pages 325-334

Ultrafiltration membrane for degumming of crude palm oil-isopropanol mixture (Article) ([Open Access](#))

Aryanti, N.^{a,b} , Wardhani, D.H.^a, Nafiunisa, A.^a 

^aDepartment of Chemical Engineering, Diponegoro University, Indonesia

^bMembrane Research Centre (MeR-C), Diponegoro University, Indonesia


Abstract

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Ultrafiltration (UF) is a membrane technology that has been applied for crude palm oil (CPO) degumming. It is considered as an alternative for the conventional CPO degumming technology because of its lower energy consumption, no need for the addition of chemicals, and almost no loss of natural oil. In this research, we separated a CPO-isopropanol mixture via laboratory-made flat-sheet polyethersulfone (PES) UF. Flux profiles confirmed that the increase in the CPO concentration resulted in lower fluxes. However, increasing the temperature from 30 °C to 45 °C initially raised the flux, but it was further decreased when the feed temperature was raised from 40 °C to 45 °C. Using UF of the CPO-isopropanol mixture at crude oil concentrations of 30 % and 40 %, we were able to reject more than 99 % phospholipids and nearly 93 % phospholipids, respectively. However, the separation of free fatty acids using this process was ineffective due to the small size of free fatty acids. Through the evaluation of the blocking mechanism in the Hermia model, it was proposed that the standard and intermediate blocking were the dominant mechanisms of filtration of CPO at a concentration of 30 and 40 %, and 50 and 60 %, respectively. © 2018 Assoc. of Chemists and Chemical Engineers of Croatia. All rights reserved.

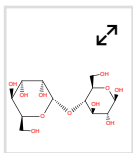
SciVal Topic Prominence

Topic: Nanofiltration | Membranes | OSN membranes

Prominence percentile: 97.950 

Chemistry database information

Substances



Author keywords

Crude palm oil Degumming Ultrafiltration

Indexed keywords

Engineering controlled terms:

Crude oil Degumming Energy utilization Fatty acids Membrane technology Mixtures
Phospholipids Ultrafiltration

Engineering uncontrolled terms

Blocking mechanisms Crude palm oil Dominant mechanism Feed temperature
Isopropanol mixtures Oil concentration Polyethersulfones Ultra-filtration membranes

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(2017) *Journal of Physics: Conference Series*

Fouling of polymeric membranes during degumming of crude sunflower and soybean oil

Pagliari, C. , Mattea, M. , Ochoa, N.
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Deacidification of soybean oil combining solvent extraction and membrane technology

Fornasero, M.L. , Marenchino, R.N. , Pagliero, C.L.
(2013) *Advances in Materials Science and Engineering*

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Funding sponsor	Funding number	Acronym
Ministry of Science Research and Technology		
Foundation for Fundamental Research on Matter		

Funding text

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The initiative to start with publication of a new journal Chemical and Biochemical Engineering Quarterly (CABEQ) as a part of European Alpe-Adria initiative for regional cooperation is described. Given are the main goals for the new journal published by Croatian Association of Chemical Engineers, formation of the international editorial board, and experiences gained during 15 years of publishing. Presented are statistical data on published papers, country of origin of authors, and classification of papers into chemical and biochemical engineering fields. The experiences gained during the last two years with publishing on Internet and planned activities in future are also presented. A view on validation of journals by ISI impact factor is discussed.

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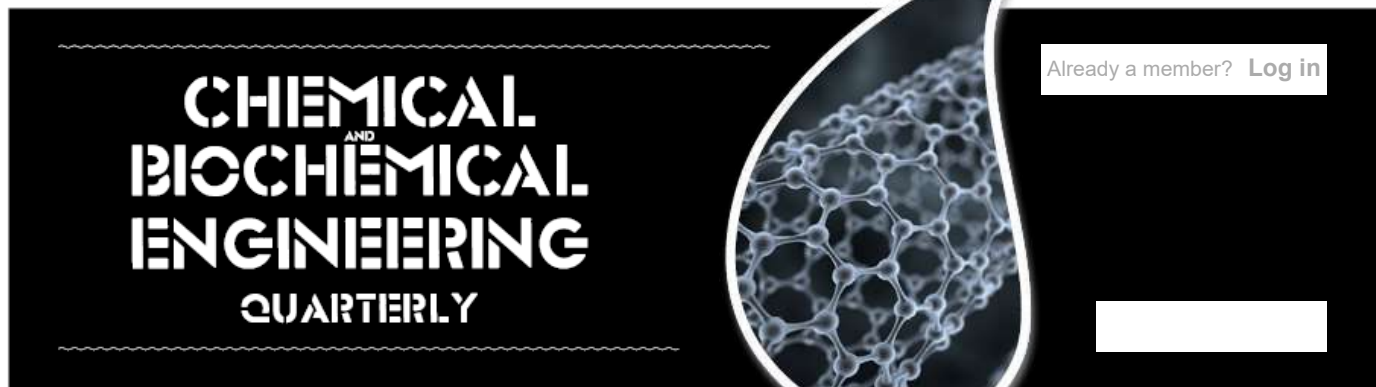
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Intensification of the Rate of Diffusion-controlled Electrochemical and Catalytic Reactions at a Helical Coil by a Fixed Bed Turbulence Promoter



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M. A. El-Naggar,^{a,*} M. S. Mansour,^a A. H. El-Shazly,^{a,b} S. A. Nosier,^a Y. A. El-Taweel,^a and G. H. Sedahmed^a

^aChemical Engineering Department, Alexandria University, Alexandria, Egypt

^bChemical and Petrochemicals Engineering Department, Egypt-Japan University of Science and Technology, El-Arab City, Alexandria, Egypt

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Intensifications of the rate of diffusion-controlled electrochemical and catalytic reactions taking place at the outer surface of a helical coil imbedded in an inert fixed bed of cylinders acting as turbulence promoters was studied using the electrochemical technique. The technique involved measuring the limiting current of the cathodic reduction of $K_3Fe(CN)_6$ in a large excess of NaOH as a supporting electrolyte.

The variables studied were solution velocity, packing cylinder diameter, helical coil tube diameter, and physical properties of the solution. The rate of mass transfer at the outer surface of the coil was found to increase with increasing solution velocity and decreasing packing cylinder diameter, helical coil tube diameter was found to have a little effect on the rate of mass transfer. The data were correlated with the dimensionless equation

$$Sh = 7.14 \cdot 10^{-4} Sc^{0.33} Re^{0.52} \left(\frac{d_p}{d} \right)^{-2.32}$$

Implication of the above equation for the design and operation of the present reactor was pointed out. The potential importance of using the inner surface of the coil as a built-in heat transfer facility for conducting exothermic electrochemical and catalytic reactions requiring rapid cooling was highlighted. The possibility of using multiple imbedded coaxial helical coils of different coil diameter in practical catalytic reactors to increase their rate of production was noted.

Keywords:

mass transfer, heat transfer, helical coils, fixed beds, turbulence promoters

Introduction

Recently, there has been substantial progress in the design of heterogeneous reactors (electrochemical reactors and catalytic reactors) used to conduct exothermic catalytic and electrochemical reactions, especially those involving heat-sensitive material that need rapid cooling. Previous studies dealing with such reactions have used heat exchanger/reactor, which consists of an array of vertical or horizontal tubes; the outer surface of the tubes acts as a catalyst support or electrode, while the inner surface of the tubes acts as a heat exchanger through which a coolant is passed to absorb excess heat generated on the outside surface^{1–4}. Such reactors with a built-in heat transfer facility would reduce the capital costs of the process owing to the elimination of the need for an external heat exchanger. Exam-

ples of exothermic reactions that need rapid temperature control include electro-organic synthesis and immobilized cell biochemical reactions. According to Baily and Ollis⁵, cells use chemical energy quite efficiently; like any real process, some of the energy in the substrate is released as heat. This metabolic heat generation dictates cooling requirements for bioreactors which harbor the cells. During electro-organic synthesis, a considerable amount of heat (Q) is generated as a result of the conversion of electrical energy used to overcome polarization at the two electrodes and ohmic drop in the solution into heat, according to the equation^{6,7}

$$Q = I \left(\eta_{\text{cathode}} + \eta_{\text{anode}} + IR - \frac{T\Delta S}{zF} \right). \quad (1)$$

This heat may adversely affect reaction selectivity and product yield besides decomposing heat-sensitive products. In case of exothermic cata-

*Corresponding author, e-mail: m.abbas.elnaggar@alexu.edu.eg

Inhibition of Copper Corrosion in NaCl Solution by Propolis Extract



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V. Grudić, I. Bošković,* and A. Gezović

Faculty of Metallurgy and Technology,
University of Montenegro, Džordža Vašingtona BB,
81000 Podgorica, Podgorica, Montenegro

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This study investigates the possibility of corrosion inhibition of copper by propolis extract in 0.51 mol dm⁻³ NaCl solution. The influence of propolis extract concentration and temperature on corrosion behaviour of copper was studied using electrochemical methods. The protective effect of propolis coating deposited on the electrode surface and dried in air was also examined. The results of the study showed that with increasing propolis extract concentration in the solution there was a slight increase in open circuit copper potential to positive values, an increase in polarization resistance, and a decrease in the corrosion current density. A significantly more prominent change of all three of these parameters was observed in the case of propolis coating. Potentiodynamic measurements indicate that propolis extract acts as a mixed corrosion inhibitor, which is adsorbed on the electrode surface according to Langmuir's isotherm. The decrease in inhibition efficiency with increasing temperature indicates physical adsorption. Adsorption of propolis on the electrode surface was confirmed by the spectrophotometric method.

Keywords:

copper, propolis, corrosion, adsorption

Introduction

Because of its very good constructional properties, high electrical and thermal conductivity, and its relatively noble properties, copper is widely used in various industries. Research has shown that copper corrodes in chlorine-containing media^{1–3}. The corrosive effect of aggressive components in electrolytes in practice can be reduced in many ways, but most often with the use of inhibitors. The use of environmentally harmful chemicals, including many effective corrosion inhibitors, has been reduced drastically in recent years. Therefore, many alternative natural products, eco-friendly or green corrosion inhibitors have been developed⁴. Apart from herbal extracts, honey and propolis are natural products, which can also be used as corrosion inhibitor for metals and alloys. Previous research has shown that ethanol extract of propolis significantly inhibits corrosion of steel^{5–7} and copper alloys⁸ in sulphate and chloride solutions.

Propolis is a resinous substance, the composition of which depends on vegetation, time, and collection area. The colour of this substance varies from yellow green to dark brown, depending on the source and age. It is accepted that propolis consists of 50 % resin, 30 % bees wax, 10 % essential and

aromatic oils, 5 % pollen, and 5 % other substances, usually flavonoids^{9,10}. In addition, propolis contains minerals, such as iron and zinc, and vitamins A, B1, B2, B6, C, and E. Apart from medical purposes, propolis is also used in industry. It is considered that some organic substances in propolis capable of inhibiting corrosion. Namely, flavonoids are substances with very good antioxidative properties. The following flavonoids have been found in propolis: galangin, quercetin, kempferol, apigenin, pinocembrin, and pinobanksin. All of them contain polyphenol and phenol groups (Scheme 1). Owing to such composition, the inhibition effect of propolis is based on adsorption and formation of a protective layer on the surface of the metal.

The aim of this work was to investigate the inhibition effect of propolis extract as a new and non-toxic inhibitor of copper corrosion in a 0.51 mol dm⁻³ NaCl solution. The experiments were performed by electrochemical methods and UV-VIS spectroscopy.

Experimental

Materials preparation

Cylindrical copper samples (purity 99.99 %) were soldered beforehand to insulated copper wires in order to achieve good electrical contact, and in-

*Corresponding author: Ivana Bošković, E-mail: ivabo@ac.me

Drug Mass Transfer Mechanism, Thermodynamics, and *In Vitro* Release Kinetics of Antioxidant-encapsulated Zeolite Microparticles as a Drug Carrier System



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Z. Lyubenova Yaneva,^{a,*} N. Valkanova Georgieva,^a
L. Lyubomirova Bekirska,^a and S. Lavrova^b

^aChemistry Unit, Department of Pharmacology, Animal Physiology and Physiological Chemistry, Faculty of Veterinary Medicine, **Trakia University**, Students Campus, 6000 Stara Zagora, **Bulgaria**

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^bFaculty of Chemical System Engineering, University of Chemical Technology and Metallurgy, 8, Kliment Ohridski Blvd., Sofia, Bulgaria

The aim of the present study was to develop a new vitamin E-zeolite drug carrier system, and investigate the mass transfer mechanism of the antioxidant encapsulation and release on/from the mineral matrix by thermodynamic and kinetics sorption/desorption experiments and mathematical modelling of the experimental data. The surface, morphological and spectral characteristics of the vitamin and the zeolite were determined by Boehm titration, SEM, FTIR and UV/Vis spectrophotometric analyses. Intraparticle diffusion was not the only rate-limiting mechanism, as the mixed-order kinetics model gave the highest regression coefficient (R^2) and lowest SSE, MSE, RMSE, and AIC_c values. The thermodynamic study confirmed the endothermic nature of the spontaneous encapsulation process and increased degrees of randomness at the solid-liquid interface. The *in vitro* release results were best modelled by the zero-order and sigmoidal models. The results obtained are essential for the development of innovative vitamin E-carrier systems for application in human and veterinary medicine.

Keywords:

vitamin E, zeolite, drug encapsulation/release, kinetics, thermodynamics, mathematical modelling

Introduction

All tocopherols and tocotrienols are nonpolar and present mainly in the lipid phase. Vitamin E is the main dietary fat-soluble antioxidant, playing important roles in the body. It is a family of four tocopherols (α , β , γ and δ) and four corresponding tocotrienols (α , β , γ and δ), of which α -tocopherol has the highest biological activity. Vitamin E acts as a chain-breaking antioxidant preventing the propagation of free radical reactions, and thus consumption of vitamin E has been widely considered to help reduce risk of many chronic diseases, such as cardiovascular diseases.^{1,2}

Besides protecting them from the harsh processing conditions and adverse storage environment, the encapsulation of bioactive compounds can also achieve targeted delivery and controlled release of entrapped nutrients to the specific site.²

Porous materials such as zeolites, zeolitic materials, activated carbons, silica gels, and metal oxides are of tremendous importance in many areas of research and development, mainly in adsorption, catalysis, energy storage, electrochemistry, and biomedical engineering.^{3–5}

However, the performance of these materials in the adsorption of organic macromolecules is not optimal because of disordered pore structure, low specific pore volume, and significant amount of micropores, which only allows the adsorption of relatively small molecules. Among these porous materials, natural, modified and synthetic zeolites have been extensively used in various industrial adsorption and separation processes due to their ordered internal structure and the availability of mesopores.^{3,6}

Adsorption of vitamin E on solid surfaces has attracted significant attention due to its importance for the pharmaceutical industry and medicinal field.^{7–10} Recently, Kavalenko and Kuznetsova reported the adsorption of vitamin E on carbon-con-

*Corresponding author: Tel.: +359 898399203, E-mail: z.yaneva@abv.bg